THE PHOTOLUMINESCENCE SPECTRA OF N+, Zn+ IMPLANTED GaAs 1-xPx AT 1.8-4.2K

Xu Jun-ying, Chen Liang-hui, Gong Ji-shu, Xu Zhong-ying, Li Yu-zhang, Zhuang Wei-hua, Xu Jing-zong and Wu Ling-xu

Institute of Semiconductors, Academia Sinica, Beijing, China

The photoluminescent behavior of N<sup>+</sup> and Zn<sup>+</sup> implanted GaAs<sub>1-x</sub>P<sub>x</sub> has been investigated at 1.8-4.2K. The experimental results show that the allow composition x, at which the conversion of N-Zn transition N and bound excit. onic recombination occurs, depends on the nitrogen and zinc doping concen-

## 1. INTRODUCTION

Previously Campbell et al. studied the N,Zn photoluminescence (PL) behavior in their N+, Zn+ GaAs<sub>1-x</sub>P<sub>x</sub> samples at 77K and observed that the conversion of the transition involving an electron bound to a nitrogen isoelectronic trap and a hole bound to a Zn acceptor (referred to as N-Zn transition) to nitrogen bound excitonic recombination at x=0.65. They presented a localized model to explain

Motivated by the influence of different N,Zn concentrations of luminescent behavior, we found the composition x, at which the conversion of N-Zn transition to N bound excitonic recombination occurs, (denoted as X(N-Zn)-N depends on N.A. concentration, the more the concentration of N,Zn the higher the X(N-Zn)-N. W have calculated the ratio of r of the transition probabilities between N-Zn transition sition and N bound excitonic recombination of three different N, Zn concentration the results are in fair agreement with our experimental results.

## 2. EXPERIMENTAL RESULTS AND ANALYSIS

The GaAs<sub>1-x</sub>P<sub>x</sub> samples used in this work were grown by conventional VPE technique. Their characteristics are summarized in Table I. Figure 1 shows typical PL spectra of  $GaAs_{1-x}P_x$  (x=0.43) samples at 4.2K. In Figure 1(a) the main per (1.923eV) was identified nitrogen bound excitonic recombination (N) while in Figure 1(b,c,d) we observed N-Zn transitions with energies of 1.922eV, 1.900 and 1.898eV respectively.

Figure 2 shows the PL spectra of  $GaAs_{1-x}P_X(x=0.49)$  at 4.2K. The related emissions were 1.955eV, 1.953eV, 1.945eV respectively for a,b and c. From Fig.1 and Fig.2 it is clear that for the samples with same x, the more Zn+ was implanted the more the N-Zn peak energy decreased, and for samples with different composition but with same N+, Zn+ concentrations, the higher the x value was, the less the N-Zn peak energy decreased.

Figure 3 shows PL emission peak energies vs. composition x. Because of different implantation energies and annealing conditions, the Zn concentration of group III samples is estimated to be 1019 cm<sup>-3</sup> while group II samples 10<sup>18</sup> cm<sup>-3</sup>. Concentrations for all cases are approximately 10<sup>18</sup> cm<sup>-3</sup> 2. By comparing the emission spectra of Nt. Gales. spectra of N<sup>+</sup>:  $GaAs_{1-x}P_x$  with these of N<sup>+</sup>,  $Zn^+$ :  $GaAs_{1-x}P_x$ , we found that the  $Zn^+$  acceptor causes the spectra of N<sup>+</sup>: $GaAs_{1-x}P_x$  to shift to lower energies. It is attributed to the N-Zn transition. The magnitude of this spectral shift is found to be composition x dependent, decreasing with x. We believe it is an indication of gradual conversion from N-Zn transition to N transition. We also notice that with the increase of Zn concentration the spectral shift tends to be less x depe dent, that is the complete conversion from N-Zn transition to N transition will occur at high x value. Our experimental results can be well understood using localized model presented by J.C. Campbell et al.! In the direct gap region, the wavefunction of Zn-bound hole is sufficiently delocalized so that its overlaption to the content of the with the wavefunction of the trapped electron exceeds that of bound excitonic

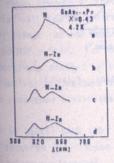
The N-Zn transition becomes dominant. With the increase of x, the Bohn radius The N-Zn transition becomes smaller, the overlap of the wavefunction of Zn bound fole becomes trapped electron decreases. On the other trapped that of the N trapped electron decreases. of Zn bound hole decomes, the overlap of the wavefunction of Zn bound note and that of the N trapped electron decreases. On the other hand, with the note and that of X, the Bohr radius of N-bound exciton gets smaller than the note of X, the Bohr radius of N-bound exciton gets smaller than the note of X, the Bohr radius of N-bound exciton gets smaller than the note of X, the Bohr radius of N-bound exciton gets smaller than the note of X, the Bohr radius of N-bound exciton gets smaller than the note of X, the Bohr radius of N-bound exciton gets smaller than the note of X, the Bohr radius of N-bound exciton gets smaller than the note of X, the no note and that of the Bohr radius of N-bound exciton gets smaller, the N bound hole increase of x, the Bohr radius of the nitrogen trap and N bound hole increase of x, the bound in the vicinity of the nitrogen trap and N bound hole becomes more localized in the vicinity of the nitrogen trap and N bound excitonic becomes more localized. As a result of these two effects the N-Zn transition is gradually localized to the N recombination and finally dominate the spectrum of the spectrum. hole increases. As a combination and finally dominate the spectrum of N,Zn: GaASI-XPX Numerically we have calculated the probability ratio r between Zn bound hole

Numerically we have and N hound hole appearing at N center at three different appearing at N center at three different, appearing at N center at three different, concentrations (Fig. 4). It is clear that with the increase of x, r becomes N.7n concentrations, implying the conversion of N-Zn transition to N transition. with the increase of N,Zn concentration the decrease of r tends to slow down, with the increase of the conversion on N,Zn concentrations. gesting the dependency observed the N-Zn transition and N bound excitonic we have simultaneously observed the N-Zn transition and N bound excitonic

We have simultaneously specified the H-ZH transition and N bound excitonic emission in N<sup>1</sup>,Zn<sup>+</sup> GaAs<sub>1-x</sub>P<sub>x</sub>(x=0.57) at 1.8K and 4.2K (Fig.5). It is the first observation of the competitive process between these two transitions.

TABLE I : The characteristics of samples used in experiments

|           | implanted element | ion<br>energy<br>(KeV) | ion<br>influence<br>(Cm <sup>-2</sup> )                        | annealing<br>temp.<br>(C") | annealing<br>duration |
|-----------|-------------------|------------------------|--|----------------------------|-----------------------|
| 7         | N <sup>+</sup>    | 170                    | 3-4×10 <sup>14</sup>   | 900                        | 1 hr.                 |
| group II  | N <sup>+</sup>    | 170                    | 3-4×10 <sup>14</sup>   | 900                        | 1 hr.                 |
|           | Zn+               | 170                    | 5×10 <sup>13</sup><br>5×10 <sup>14</sup><br>5×10 <sup>15</sup> |                            |                       |
| group III | N <sup>+</sup>    | 160                    | 1×10 <sup>14</sup>   | 850                        | 35 min.               |
|           | Zn <sup>+</sup>   | 80                     | 4×10 <sup>15</sup>   |                            |                       |



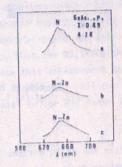


Fig. 1: The PL spectra of GaAsP(x=0.43) at 4.2K: (a) spectrum from a sample of group I, with N implantation only; (b),(c) and (d) spectra from samples of group II, with N influence of 3-4x1014<sub>cm</sub><sup>-2</sup> and Zn influences of 5x1013, 5x1014 and 5x1015<sub>cm</sub><sup>-2</sup>, respectively.

FIG. 2: The spectra of GaAsP(x=0.49)at 4.2K: (a) spectrum from a sample of group I, with N implantation only; (b) and (c) spectra from samples of group II with N influence of 3-4x10<sup>14</sup> and 5x10<sup>15</sup>cm<sup>-2</sup>, respectively.

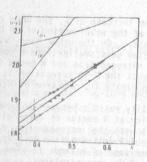
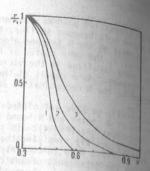


FIG. 3: Composition dependence of PL spectral peak energy of  $Zn^+N^+$  ion implanted  $GaAs_{1-x}P_x$  at 4.2K: curve 1 for group I samples, curve 2  $n_{2n-1}x_10_18c_m-3$  3.  $n_{2n-1}x_10_18c_m-3$  3.  $n_{2n-1}x_10_18c_m-3$  3.  $n_{2n-1}x_10_18c_m-3$  3.  $n_{2n-1}x_10_18c_m-3$  3. The ratio was normalized to the ratio  $n_{2n-1}x_10_18c_m-3$ 



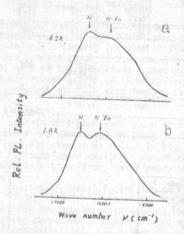


FIG. 5: PL spectra of N<sup>+</sup>,Zn<sup>+</sup>:  $GaAs_{1-x}P_{\chi}(x=0.57)$  at (a) 4.2K and (b) 1.8K.

## REFERENCES

1. J.C. Campbell, N. Holonyak, Jr., and M.H. Lee, Phys. Rev. B10 (1974) 1755-2. K.H. Lee, W.H. Zhuang, Z.G. Lee, D.M. Zhou, Jnl. of Beijing Normal Univ.