

## Crystal Structures of Tungsten Disulfide and Diselenide\*

W. J. SCHUTTE, J. L. DE BOER, AND F. JELLINEK

*Laboratory of Inorganic Chemistry, Materials Science Center of the University, Nijenborgh 16, 9747 AG Groningen, The Netherlands*

Received December 29, 1986

The crystal structures of the 2*H*- and 3*R*-forms of WS<sub>2</sub> and of WSe<sub>2</sub> have been refined from single-crystal data. The results are summarized and the interatomic distances are compared with those in related compounds. © 1987 Academic Press, Inc.

### Introduction

It has been known for many years that tungsten disulfide, WS<sub>2</sub> (1), and diselenide, WSe<sub>2</sub> (2), have layer structures isotypic with hexagonal MoS<sub>2</sub> (3, 4). In addition to the common hexagonal 2*H*-form of WS<sub>2</sub> a rhombohedral form, 3*R*-WS<sub>2</sub>, has also been reported (5), which is isotypic with the rhombohedral form of MoS<sub>2</sub> (6, 7). However, apart from a rough determination of the atomic positions in 2*H*-WS<sub>2</sub> and WSe<sub>2</sub> from powder X-ray diffraction data (8), no refinement of the structures of the tungsten compounds has been reported so far. We have undertaken such a study in connection with calculations of the electronic band structures of these compounds (9), for which the atomic positions had to be accurately known.

### Experimental

Single crystals of the tungsten dichalcogenides had been prepared by J. C. Wil-

dervanck (10) by means of chemical transport using chlorine (2*H*-WS<sub>2</sub>) or bromine (3*R*-WS<sub>2</sub>, WSe<sub>2</sub>) as transport agent. Very thin plate-like crystals were investigated by X-ray diffraction using a Nonius CAD4 diffractometer; monochromatized MoK $\alpha$  radiation ( $\lambda = 0.71071 \text{ \AA}$ ) was employed. Unit cell parameters (Table I) were determined by least-squares fits of the optimized setting angles of about 20 reflections in the  $\theta$  ranges 20–24° (2*H*-WS<sub>2</sub>), 30–35° (WSe<sub>2</sub>), and 27–31°, respectively; the parameters agree with those given by previous authors (5, 11).

Intensity data were collected in hemispheres up to  $\theta$  values of 50° (2*H*-WS<sub>2</sub>), 45° (3*R*-WS<sub>2</sub>), and 35° (WSe<sub>2</sub>), respectively. A modified version (12) of the CAD4 program was used to measure each reflection of the thin plate-shaped crystals in its position of minimal absorption; no absorption corrections were applied. The intensities of equivalent reflections were averaged and corrected for Lorentz and polarization effects; reflections with  $I < 2.5 \sigma(I)$  were discarded. The structure factors of the remaining reflections (see Table I) were used

\* Dedicated to Dr. H. Nowotny.

TABLE I  
CRYSTALLOGRAPHIC DATA OF  $\text{WS}_2$  AND  $\text{WSe}_2$   
(STANDARD DEVIATIONS ARE  
GIVEN IN PARENTHESES)

	$2H\text{-WS}_2$	$\text{WSe}_2$	$3R\text{-WS}_2$
Space group	$P6_3/mmc$	$P6_3/mmc$	$R3m$
$a$ (Å)	3.1532(4)	3.282(1)	3.158(1)
$c$ (Å)	12.323(5)	12.96(1)	18.49(1)
$c/a$	3.908	3.949	5.855
$V$ (Å <sup>3</sup> )	106.1	120.9	159.6
$Z$	2	2	3
Independent reflections	245	119	205
$z$ (S, Se)	0.6225(6)	0.6211(4)	<sup>a</sup>
$R_F$ (%)	6.4	6.9	4.5

<sup>a</sup>  $z(\text{W}) = 0.0000(5)$ ;  $z(\text{S I}) = 0.2497(6)$ ;  $z(\text{S II}) = 0.4190(7)$ .

in the refinements by a full-matrix least-squares procedure. The atomic scattering factors were those of the XTAL system (13); anomalous dispersion factors were taken from the "International Tables for X-Ray Crystallography" (14). The final values of  $R_F$  are included in Table I.

## Results and Discussion

It was confirmed that  $2H\text{-WS}_2$  and  $\text{WSe}_2$  are isotypic with  $2H\text{-MoS}_2$  (3, 4), with space group  $P6_3/mmc$ . The tungsten atoms lie in  $2(c)$ :  $\pm (1/3, 2/3, 1/4)$ , and the chalcogen atoms in  $4(f)$ :  $\pm (1/3, 2/3, z; 1/3, 2/3, 1/2-z)$ . The final values of  $z$  are included in Table I and interatomic distances in Table II. The chalcogen parameter derived by Kalikham (8) for  $\text{WSe}_2$  ( $z \approx 0.620$  in our setting) agrees with our results, but his value for  $2H\text{-WS}_2$  ( $z \approx 0.614$ ) is rather far off the mark. This is understandable because of the relatively small scattering factor of sulfur relative to tungsten.

Just as  $3R\text{-MoS}_2$  (6), the  $3R$ -form of  $\text{WS}_2$  crystallizes in space group  $R3m$  with tungsten and two independent sets of sulfur atoms all in positions  $3(a)$ :  $(00z; 2/3, 1/3, 1/3 + z; 1/3, 2/3, 2/3 + z)$ . The final parameters

TABLE II  
INTERATOMIC DISTANCES (Å) IN  $\text{WX}_2$  ( $X = \text{S, Se}$ )  
AND THEIR STANDARD DEVIATIONS  
(IN PARENTHESES)

	$2H\text{-WS}_2$	$\text{WSe}_2$	$3R\text{-WS}_2$
Within $\text{WX}_2$ layers			
W-3X	2.405(5)	2.526(4)	2.39(1)
W-3X	2.405(5)	2.526(4)	2.42(1)
X-1X	3.14(2)	3.34(1)	3.13(2)
X-6X = W-6W	3.153	3.282	3.158
Between $\text{WX}_2$ layers			
X-3X	3.53(1)	3.67(1)	3.54(1)

(and their standard deviations) are  $z(\text{W}) = 0.0000(5)$ ;  $z(\text{S I}) = 0.2497(6)$ ;  $z(\text{S II}) = 0.4190(7)$ ; interatomic distances are listed in Table II.

From Table II it is seen that the trigonal prisms formed by the chalcogen atoms around a tungsten atom are fairly regular, the prism edges parallel to  $c$ ,  $X\text{-}1X$  ( $X = \text{S, Se}$ ), being of about the same length or slightly longer than the edges  $X\text{-}6X$  perpendicular to  $c$ . The same is true of  $\text{MoX}_2$  (4, 7), i.e.  $\text{MoTe}_2$  (15, 16), and also  $\text{ZrCl}_2$  (17); in all these phases the metal has a  $d^2$ -configuration. In contrast, in  $\text{NbS}_2$  (18),  $\text{NbSe}_2$  (19, 20), and  $\text{TaSe}_2$  (21), where the metal has  $d^1$ -configuration, the prism edges parallel to  $c$  are shorter than those perpendicular to  $c$ . It may be remarked that trigonal-prismatic coordination is expected to be stable only for metals with a  $d^0$ ,  $d^1$ , or  $nd^2$  ( $n > 3$ ) configuration (22). In the semi-conducting molybdenum and tungsten dichalcogenides the distances  $X\text{-}3X$  between the layers are considerably longer than the  $X\text{-}X$  distances within the layers; this difference is smaller in the metallic niobium and tantalum dichalcogenides.

## Acknowledgment

Financial support by The Netherlands Organization for the Advancement of Pure Research (ZWO) through The Netherlands Foundation for Chemical Research (SON) is gratefully acknowledged.

## References

1. A. E. VAN ARKEL, *Rec. Trav. Chim. Pays-Bas* **45**, 437 (1926).
2. O. GLEMSER, H. SAUER, AND P. KÖNIG, *Z. Anorg. Allg. Chem.* **257**, 241 (1948).
3. R. G. DICKINSON AND L. PAULING, *J. Amer. Chem. Soc.* **45**, 1466 (1923).
4. K. D. BRONSEMA, J. L. DE BOER, AND F. JELLINEK, *Z. Anorg. Allg. Chem.* **540/541**, 15–17 (1986).
5. J. C. WILDERVANCK AND F. JELLINEK, *Z. Anorg. Allg. Chem.* **328**, 309 (1964).
6. F. JELLINEK, G. BRAUER, AND H. MÜLLER, *Nature (London)* **185**, 376 (1960).
7. Y. TAKÉUCHI AND W. NOWACKI, *Schweiz. Mineral. Petrogr. Mitt.* **44**, 105 (1964).
8. V. L. KALIKHMAN, *Izv. Akad. Nauk SSSR, Neorg. Mater.* **19**, 1060 (1983); *Inorg. Mater.* **19**, 957 (1983).
9. R. COELHOORN, dissertation, Groningen (1985).
10. J. C. WILDERVANCK, dissertation, Groningen (1970).
11. B. F. MENTZEN AND M. J. SIENKO, *Inorg. Chem.* **15**, 2198 (1976) and literature cited therein.
12. J. L. DE BOER AND A. J. M. DUISENBERG, *Acta Crystallogr. A* **40**, C410 (1984).
13. J. M. STEWART AND S. R. HALL Eds., "The XTAL System of Crystallographic Programs," 2nd ed., Univ. of Maryland, College Park, (1985).
14. "International Tables for X-Ray Crystallography," Vol. IV, Kynoch Press, Birmingham (1974).
15. D. PUOTINEN AND R. E. NEWNHAM, *Acta Crystallogr.* **14**, 691 (1961).
16. O. KNOP AND R. D. MACDONALD, *Canad. J. Chem.* **39**, 897 (1961).
17. S. I. TROYANOV AND V. I. TSIRELNIKOV, *Vestn. Mosk. Univ. Khim.* **14**, 67 (1963).
18. B. MOROSIN, *Acta Crystallogr. B* **30**, 551 (1974).
19. B. E. BROWN AND D. J. BEERNTSEN, *Acta Crystallogr.* **18**, 31 (1965).
20. M. MAREZIO, P. D. DERNIER, A. MENTH, AND G. W. HULL, JR., *J. Solid State Chem.* **4**, 425 (1972).
21. R. HUISMAN, dissertation, Groningen (1969).
22. R. HUISMAN, R. DE JONGE, C. HAAS, AND F. JELLINEK, *J. Solid State Chem.* **3**, 56 (1971).